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# Synthesis and Mass Spectrometric Analysis of Cyclostellettamines H, I, K and L

# Achim Grube, [a] Christoph Timm, [a] and Matthias Köck\*[a]

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Very recently the new cyclostellettamines H, I, K and L were identified from a Brazilian sponge of the genus Pachychalina. They were isolated together with the known cyclostellettamines A-G in a mixture of only 1.5 mg. To obtain further material for biological investigations, the synthesis of the four new cyclostellettamines has been carried out. Since mass spec-

trometry plays an essential role in identifying these compounds a systematic analysis of the cyclostellettamines is discussed.

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#### Introduction

Piperidine and pyridine (pyridinium) alkaloids are widely distributed in marine sponges of different genera.<sup>[1]</sup> The order Haplosclerida (genera Haliclona, Xestospongia and Amphimedon) has proved to be a particularly rich source of alkaloids containing 1,3-dialkylpyridine or piperidine motifs, [2] many of which have quite complex structures. Members of this alkaloid class with a lower structural complexity are the halitoxins, [3] the haliclamines [4] and the cyclostellettamines.<sup>[5]</sup> The first six members of the cyclostellettamines (A to F) were reported by Fusetani and co-workers in 1994.<sup>[5a]</sup> They differ in the lengths of the alkyl chains – the chain length varies from 12 to 14 carbon atoms - that connect the two pyridinium rings in positions 1 and 3. Very recently, we have identified eleven cyclostellettamines from a sample of a Brazilian sponge of the genus Pachychalina, of which cyclostellettamines A–G (2)<sup>[6]</sup> were already known. <sup>[5c]</sup> The chain lengths of the new cyclostellettamines H (1), I (3), K (4) and L (5) were identified as 10 to 14.<sup>[7]</sup> Since it was not possible to get pure samples of the new compounds a synthetic approach was carried out. Furthermore, the as vet unknown cyclostellettamine O (6) is also discussed in this manuscript.<sup>[8]</sup> Cyclostellettamines I (3), K (4) and L (5) are the first members of this family in which the chain length of the two alkyl chains differs by more than two carbon atoms (Figure 1). A difference of three in the chain lengths is known for the structurally related haliclamines (tetrahydropyridine derivatives of the cyclostellettamines).[4a]

Fax: +49-471-48311425

E-mail: mkoeck@awi-bremerhaven.de

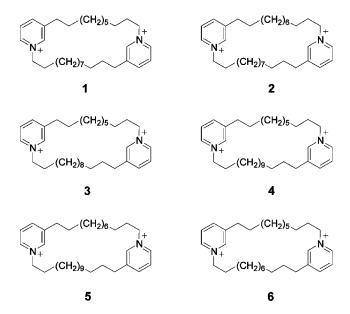


Figure 1. Cyclostellettamines H (1), G (2), I (3), K (4), L (5) and Q (6).

## **Methods and Results**

Cyclostellettamines G (2), H (1), I (3), K (4), L (5) and Q (6) were synthesised following a procedure developed by Baldwin et al.<sup>[9]</sup> Depending on the commercial availability of the starting material, pyridyl alcohol monomers 12 were prepared in three to five steps, starting either from the diacid  $(C_{14})$ , the diol  $(C_{10},\,C_{11},\,C_{13})$  or the bromo alcohol (C<sub>12</sub>). The diacid **7a** was reduced to the diol **8a** by treatment with BH<sub>3</sub>·SMe<sub>2</sub>. Monobromination<sup>[10]</sup> of the diols 8a,b,d,e yielded the bromo alcohols 9a,b,d,e, which were then protected as the THP ethers 10a-e and coupled with 3-picoline to give the protected 3-alkylpyridines 11a-e. Deprotection

<sup>[</sup>a] Alfred-Wegener-Institut für Polar- und Meeresforschung in der Helmholtz-Gemeinschaft, Am Handelshafen 12, 27570 Bremerhaven, Germany

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HO OH OH OH 
$$\frac{2}{n}$$
 OH  $\frac{3}{n}$  OH  $\frac{3}{n}$  OTHP  $\frac{4}{n}$  OTHP  $\frac{4}{n}$  OTHP  $\frac{5}{n}$  OTHP  $\frac{5}{n}$  OTHP  $\frac{11}{12}$  OTHP  $\frac{12}{n}$  OH  $\frac{3}{n}$  OH  $\frac{3}{n}$  OTHP  $\frac{12}{n}$  OH  $\frac{3}{n}$  OTHP  $\frac{12}{n}$  OH  $\frac{3}{n}$  OTHP  $\frac{12}{n}$  OH  $\frac{3}{n}$  OTHP  $\frac{1}{n}$  OTHP

**a**: n = 11 **b**: n = 10 **c**: n = 9 **d**: n = 8 **e**: n = 7

Figure 2. Preparation of the monomeric building block 12: 1) 4 equiv. BH<sub>3</sub>·SMe<sub>2</sub>, THF, 0 °C to room temp.; 2) 1.1 equiv. HBr, toluene,  $\Delta$ ,  $-H_2O$ ; 3) 1.5 equiv. DHP, PPTS, DCM, room temp.; 4) 2 equiv. picoline, LDA, THF, -78 °C to room temp.; 5) 2 equiv. HCl, MeOH, room temp.

with HCl gave monomers **12a–e** in 18 to 39% yield (see Figure 2).

Prior to dimerisation, one monomer was activated by converting the alcohol into a chloride with SOCl<sub>2</sub>, followed by protection of the nitrogen as the *N*-oxide by treatment with *m*-CPBA (see Figure 3). Coupling of 12 and 14 was achieved by treatment with NaI (nucleophilic addition). The resulting dimer was activated and deprotected in one step with PBr<sub>3</sub>. Cyclisation to the target compounds was performed under pseudo high dilution conditions by slowly adding the bromo dimer with a syringe pump to a refluxing solution of NaI in butan-2-one to give cyclostellettamines G (2), H (1), I (3), K (4), L (5) and Q (6) in 28 to 61% yield (not optimised) starting from monomers 12 and 14 (see Figure 4).

Figure 3. Activation of the monomeric building block **12**: 6) 1.2 equiv. SOCl<sub>2</sub>, dioxane, 0 °C–room temp.; 7) 1.1 equiv. *m*-CPBA, DCM, 0 °C to room temp.

MS and MS/MS methods are essential for the structure elucidation of cyclstellettamines since the NMR spectra of these compounds are very similar to each other. Determination of the high resolution masses of cyclostellettamines and generation of the corresponding molecular fragments was achieved with an ESI-oTOF mass spectrometer. Even with low fragmentation energy the molecules underwent mass spectral cleavage, which allowed MS/MS analysis without fragmentation with an ion trap or a triple quadru-

pole as analyzer (Figure 5). Molecular fragments of cyclostellettamines were generated in the ion-transfer unit of the orthogonal-TOF by setting a voltage difference between the capillary exit and the skimmer. In this setup API-CID (atmospheric pressure ionisation – collision induced dissociation) was carried out with a non-mass-selected ion stream, which made it necessary to work with pure compounds.

The fragmentation patterns under API-CID and CID-MS/MS conditions were compared. The results of the MS and API-CID-MS/MS analysis of cyclostellettamines H (1), G (2), I (3), K (4), L (5) and Q (6) are given in Table 1. The average mass deviation of the TOF instrument was 0.84 amu with a standard deviation of 0.49 amu (largest mass deviation was 1.9 amu).

In order to understand the fragmentation pattern of cyclostellettamines, MS<sup>n</sup> spectra of cyclostellettamine Q (6) were recorded with an ion-trap mass spectrometer (see Figure 6). Three fragmentation pathways are possible for cyclostellettamine Q (6): an onium reaction, a Hofmann fragmentation<sup>[2d]</sup> and a McLafferty rearrangement (see Figure 7). The two main fragments at m/z = 218 and 232 are both generated from the doubly charged precursor ion (m/z = 225, see Figure 6c), the mono iodide salt (m/z = 577, see Figure 6b) and from the m/z = 449 fragment (not shown) in an onium reaction. In this reaction a proton remains at the pyridine nitrogen to form a quaternary ammonium ion, while the localisation of the double bond in the alkyl chain is variable.<sup>[11]</sup>

In MS/MS spectra of the cyclostellettamines the fragments have a typical difference of 14 amu. This indicates the stepwise cleavage of the alkyl chain. The daughter ions obtained in the MS/MS spectra of the m/z = 225 fragment, i.e. m/z = 162 and 176, are doubly and singly charged, respectively. This leads to the assumption that the doubly charged  $M^{2+}$  ion also undergoes an onium reaction and splits in the alkyl chain. The doubly charged daughter ions at m/z = 197, 183 and 169 could arise from a McLafferty-type rearrangement of the m/z = 225 fragment, where a methyl group remains at the pyridine nitrogen. Duffield et al. have shown for N-butyl- and N-pentylpyrroles that a hydrogen transfer occurs from C-3 of the alkyl chain via a six-

12 + 14 
$$\xrightarrow{8.}$$
  $\xrightarrow{N}$   $\xrightarrow{N}$ 

f: m = 9 n = 7 g: m = 7 n = 10 h: m = 11 n = 7 i: m = 9 n = 8 j: m = 11 n = 8 k: m = 7 n = 8

Figure 4. Dimerisation and cyclisation of the monomeric building blocks 12 and 14: 8) 1.2 equiv. NaI, butan-2-one,  $\Delta$ ; 9) 4 equiv. PBr<sub>3</sub>, CHCl<sub>3</sub>, 0 °C to room temp.; 10) 2.2 equiv. NaI, butan-2-one,  $\Delta$ .

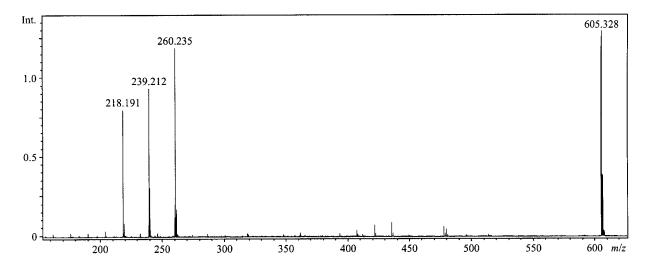


Figure 5. API-CID-MS/MS spectrum of cyclostellettamine I (3). The two daughter ions at m/z = 218 and 260 arise from the doubly charged precursor ion (m/z = 239) and the mono iodide salt (m/z = 605).

membered transition state to an  $\alpha$ -carbon of the heterocycle with a synchronous  $\alpha,\beta$ -carbon bond cleavage. [12] An analogous pathway with a following re-aromatisation could be possible for the cyclostellettamines (see Figure 8). A  $\gamma$ -cleavage that leads to a 1,2-dimethylene-1,2-dihydropyridinium ion has been described by Spiteller for alkylpyridines. [13] A 1-azoniabicyclo[4.2.0]octa-1(6),2,4-triene, which has a similar structure to the cyclostellettamines, is formed as an intermediate in this reaction (see Supporting Information). The intensity of the McLafferty fragments is lower than the fragments due to the onium reaction.

The  $[M-2I-H]^+$  fragment typically observed for the cyclostellettamines is only formed from the mono iodide salt at m/z = 577 in a so-called Hofmann fragmentation. The appearance of this Hofmann fragment seems to depend on the anion as this fragment is observed with chloride, [2d]

bromide, iodide and nitrate as anion, whereas with triflate<sup>[2d]</sup> no Hofmann fragmentation is observed. Under API-CID conditions, and only in the presence of formic acid, an  $[M-2I+H]^+$  fragment is also formed. Other acids, for example acetic acid or ascorbic acid, do not lead to the observed  $[M-2I+H]^+$  fragment. This fragment is probably generated by a reduction of the unsaturated alkyl chain with formic acid as reducing agent.

### **Discussion and Conclusion**

Since the new cyclostellettamines H, I, K and L were obtained in a 1.5 mg mixture of 11 derivatives, a synthetic approach was deemed necessary to obtain pure compounds for further biological studies. The synthetic approach intro-

Table 1. MS and MS/MS results for compounds 1-6.

		HRMS	Fragment 1	Fragment 2
1	calcd.	232.2060, C <sub>32</sub> H <sub>52</sub> N <sub>2</sub>	218.1902, C <sub>15</sub> H <sub>24</sub> N (C <sub>10</sub> )	246.2216, C <sub>17</sub> H <sub>28</sub> N (C <sub>12</sub> )
	exp.	$232.2054$ , $\Delta m = 2.7$ ppm	218.1913, $\Delta m = 4.5 \text{ ppm}$	246.2210, $\Delta m = 2.6 \text{ ppm}$
	-	FWHM: 0.0418	FWHM: 0.0399	FWHM: 0.0439
2	calcd.	239.2138, $C_{33}H_{54}N_2$	232.2060, $C_{16}H_{26}N$ ( $C_{11}$ )	246.2216, C <sub>17</sub> H <sub>28</sub> N (C <sub>12</sub> )
	exp.	239.2135, $\Delta m = 1.1 \text{ ppm}$	232.2064, $\Delta m = 1.8 \text{ ppm}$	$246.2211$ , $\Delta m = 2.0$ ppm
	-	FWHM: 0.0450	FWHM: 0.0366	FWHM: 0.0381
3	calcd.	239.2138, $C_{33}H_{54}N_2$	218.1902, $C_{15}H_{24}N$ ( $C_{10}$ )	$260.2373$ , $C_{18}H_{30}N$ ( $C_{13}$ )
	exp.	239.2136, $\Delta m = 1.0 \text{ ppm}$	218.1914, $\Delta m = 5.2 \text{ ppm}$	$260.2364$ , $\Delta m = 3.2$ ppm
	-	FWHM: 0.0457	FWHM: 0.0353	FWHM: 0.0399
4	calcd.	$246.2216$ , $C_{34}H_{56}N_2$	218.1903, $C_{15}H_{24}N$ ( $C_{10}$ )	274.2529, C <sub>18</sub> H <sub>30</sub> N (C <sub>14</sub> )
	exp.	$246.2205 \ \Delta m = 4.5 \ \text{ppm}$	218.1915, $\Delta m = 5.5 \text{ ppm}$	$274.2519$ , $\Delta m = 3.8$ ppm
	-	FWHM: 0.0430	FWHM: 0.0394	FWHM: 0.0479
5	calcd.	253.2295, $C_{35}H_{58}N_2$	232.2060, $C_{16}H_{26}N$ ( $C_{11}$ )	274.2529, C <sub>18</sub> H <sub>30</sub> N (C <sub>14</sub> )
	exp.	$253.2279$ , $\Delta m = 6.2$ ppm	232.2059, $\Delta m = 0.4 \text{ ppm}$	$274.2510$ , $\Delta m = 6.9$ ppm
	_	FWHM: 0.0462	FWHM: 0.0414	FWHM: 0.0476
6	calcd.	225.1982, $C_{31}H_{50}N_2$	218.1902, $C_{15}H_{24}N$ ( $C_{10}$ )	232.2060, $C_{16}H_{26}N$ ( $C_{11}$ )
	exp.	225.1992, $\Delta m = 4.9 \text{ ppm}$	218.1914, $\Delta m = 5.1 \text{ ppm}$	232.2061, $\Delta m = 0.4 \text{ ppm}$
	*	FWHM: 0.0339	FWHM: 0.0339	FWHM: 0.0366

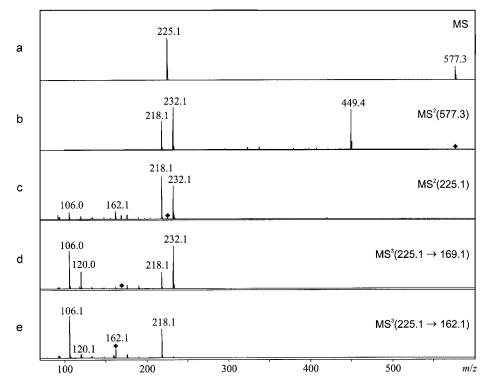


Figure 6. MS and MS<sup>n</sup> spectra of cyclostellettamine Q (6): a) MS spectrum of 6 with the doubly charged ion at m/z = 225 and the mono iodide salt (m/z = 577); b) MS/MS spectrum of the mono iodide salt as precursor ion; c) MS/MS spectrum of the doubly charged ion at m/z = 225 as precursor ion; d) MS/MS spectrum of the doubly charged ion at m/z = 169, which arises as a daughter ion from the m/z = 225 fragment; e) MS/MS spectrum of the doubly charged ion at m/z = 162, which arises as a daughter ion from the m/z = 225 fragment.

duced by Baldwin et al. proved to be an effective access to cyclostellettamines. The structure elucidation of cyclostellettamines, and 3-alkylpyridine alkaloids in general, depends strongly on mass spectrometry as the NMR spectra of these compounds are very similar to each other due to a considerable overlap of the aliphatic protons (about 50% of the proton resonances have the same resonance frequency). This investigation has described a systematic analysis of ESI-HRMS, ESI-API-CID-MS/MS and ESI-CID-MS/MS

spectra. The high resolution masses could be measured with the high accuracy required for the unambiguous identification of cyclostellettamines. The results show comparable fragmentation behaviour under ESI-API-CID-MS/MS and ESI-CID-MS/MS conditions. An advantage of API-CID-MS/MS with an oTOF is the determination of high resolution masses. Analysis of the fragments generated allows a fast and easy identification of the alkyl chain lengths of the cyclostellettamines, thus facilitating the structure elucida-

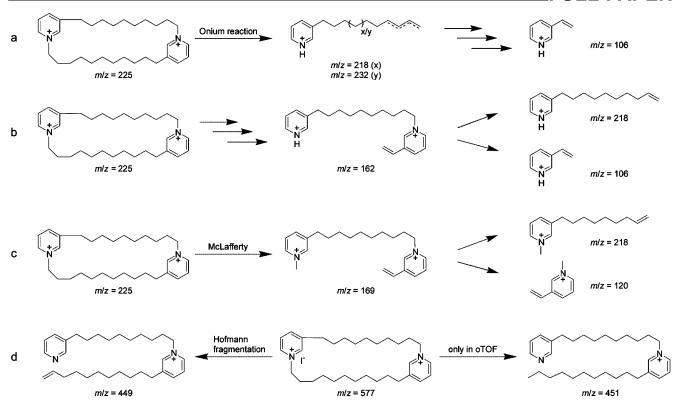


Figure 7. Proposed fragmentation pathways of cyclostellettamine Q (6) according to MS/MS analysis. The doubly charged precursor ion (m/z = 225) undergoes an onium reaction which leads to singly (a) as well as doubly (b) charged ions. Another possible fragmentation reaction of the precursor ion is the McLafferty rearrangement (c). The mono iodide salt reacts in a so-called Hofmann fragmentation and only in oTOF to a reduced form of the Hofmann fragment (d).

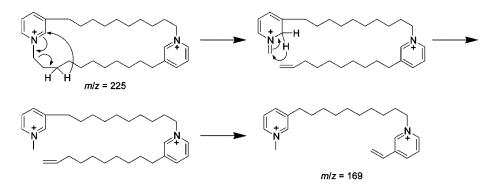


Figure 8. Possible fragmentation mechanism for cyclostellettamines related to the McLafferty rearrangement. Initially, a hydrogen transfer from C-3 of the alkyl chain via a six-membered transition state to an  $\alpha$ -carbon of the heterocycle with a synchronous  $\alpha,\beta$ -carbon bond cleavage occurs. The obtained 1-methylene-1,2-dihydropyridinium cation can undergo a re-aromatisation to form a methylpyridinium ion, which splits in the alkyl chain.

tion. Furthermore, the systematic investigation will promote a better understanding of the fragmentation patterns of these kinds of molecules in the future.

## **Experimental Section**

**General:** NMR spectra were recorded with a Bruker AM 250 (250 MHz) spectrometer. Chemical shifts are quoted in ppm and are referenced to the appropriate solvent signal. The samples were injected into a HPLC System 1100 Series (Agilent) equipped with a photodiode-array detector (Agilent) and a microTOF<sub>LC</sub> mass spectrometer (Bruker Daltonik). The detection with the DAD was

performed at a wavelength of 260 nm. For HPLC separation an XTerra RP-18 column (3.0×150 mm, 3.5 µm, Waters) was used. Separation was achieved by applying a gradient from 20% acetonitrile/80% formic acid (0.1%) to 80% acetonitrile/20% formic acid (0.1%) in 30 min and in 35 min back to the initial conditions. Total analysis time was 45 minutes with a flow rate of 0.4 mL min $^{-1}$  and an oven temperature of 30 °C. Mass spectra were acquired with a microTOF $_{\rm LC}$  mass spectrometer equipped with an ESI source (Bruker Daltonik). The following ESI inlet conditions were applied: dry gas temperature: 180 °C; dry gas flow: 10 L min $^{-1}$ ; nebulizer pressure: 1.5 bar; capillary voltage: 4500 V. For fragmentation the voltage of the capillary exit was set to 150% of the normal voltage and the voltage of skimmer 1 was set to 133% of the nor-

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mal voltage. The system was calibrated in positive mode by external calibration with sodium formiate clusters. The following reference masses were used for calibration: 226.9515, 362.9263, 430.9138, 498.9012, 566.8886, 634.8760, 702.8635 and 770.8509. Before the measurements the calibration was renewed. The standard deviation of the calibration curve (quadratic calibration) was 1.51. MS<sup>n</sup> spectra were acquired with an Esquire HCT ion trap in the positive mode equipped with an ESI source (Bruker Daltonik). FT-IR spectra were recorded on a Perkin-Elmer 1600 series spectrometer. Absorption maxima are reported in wavenumbers and the following abbreviations are used s: strong, m: medium, w: weak Elemental analysis was performed with a Heraeus CHN Rapid. Melting points were obtained with a Kofler melting point apparatus and are uncorrected. Column chromatography was performed on silica gel 60 (Merck, particle size 0.04-0.063 mm) or basic alumina. TLC was performed on aluminium plates precoated with Merck silica 60. Compounds were visualised by UV irradiation (254 nm) or dying with KMnO<sub>4</sub> solution (1% KMnO<sub>4</sub>, 6.6% K<sub>2</sub>CO<sub>3</sub>, 2% 5% NaOH solution in 100 mL of water). All solvents were purified by simple distillation, except for THF, which was distilled from sodium/benzophenone under argon. Chemicals were used as purchased. LDA solution was purchased from Fluka. The assignment of protons and carbons in the dimers is as follows:

1,13-Tridecanediol (8a): A solution of BH<sub>3</sub>·SMe<sub>2</sub> (36.83 mmol, 3.76 mL 94%) in 10 mL of THF was added within 10 min, at 0 °C, to a suspension of tridecanoic diacid (3 g, 12.27 mmol) in 40 mL of dry THF under argon in a three-necked flask equipped with a dropping funnel and a silver-coated Vigreux column. After the addition was complete and gas evolution had almost stopped the ice bath was removed, the mixture was heated to reflux and SMe2 was distilled off. When the temperature in the column head had reached 66 °C, indicating that all SMe<sub>2</sub> had been removed, the mixture was refluxed for an additional 30 min. It was then cooled down to room temperature and 30 mL of MeOH was slowly added (gas evolution). The solvent was removed and the residue was twice coevaporated with 30 mL of MeOH to give a grey solid, which was recrystallised from hexane to yield 2.2 g (83%) of 8a as white crystals. M.p. 77 °C (ref. [14] 76–77 °C). <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$ = 1.18-1.25 (br. s, 18 H,  $9 \times CH_2$ ), 1.36-1.41 (m, 4 H,  $CH_2CH_2OH$ ), 3.36 (m and t, J = 6.5 Hz, 4 H,  $CH_2OH$ ), 4.32 (t, J= 5.0 Hz, 2 H, OH; interchangeable with  $D_2O$ ) ppm. IR (KBr):  $\tilde{v}$  $= 3299 \text{ cm}^{-1} \text{ s}, 2917 \text{ s}, 2849 \text{ s}, 1462 \text{ s}, 1407 \text{ m}, 1344 \text{ m}, 1296 \text{ w}, 1257$ w, 1187 w, 1123 s, 1058 s, 1036 m, 1002 w, 958 m, 919 m, 831 w, 732 m. C<sub>13</sub>H<sub>28</sub>O<sub>2</sub> (216.21): calcd. C 72.17, H 13.04; found C 72.31, H 13.26.

13-Bromotridecan-1-ol (9a): To a suspension of tridecandiol 8a (2.2 g, 10.16 mmol) in 80 mL of toluene in a round-bottomed flask equipped with a water extractor and a cooler was added hydrobromic acid (1.7 mL 48% aqueous solution, 15.25 mmol). The mixture was refluxed for 16 h. After cooling, the solution was consecutively washed with 20 mL of 1 m HCl, 2 m NaOH, water and brine. The organic layer was dried with MgSO<sub>4</sub> and the solvent was evaporated. Flash chromatography of the residue on silica (4:1 hexane/ethyl acetate) yielded 1.53 g (54%) of 9a as a white solid.  $^1$ H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.16–1.41 (m, 18 H, 9×CH<sub>2</sub>), 1.42–1.58 (m, 2 H, CH<sub>2</sub>–CH<sub>2</sub>–OH), 1.72–1.84 (m, 2 H, CH<sub>2</sub>–CH<sub>2</sub>–Br), 3.39

(t, J = 7.0 Hz, 2 H,  $CH_2OH$ ), 3.57 (t, J = 6.5 Hz, 2 H,  $CH_2$ –Br) ppm. IR (KBr):  $\tilde{v} = 3425$  cm<sup>-1</sup> s, 2920 s, 2850 s, 1467 s, 1355 m, 1252 w, 1229 w, 1204 w, 1058 s, 970 w, 722 m, 645 s.  $C_{13}H_{27}$ BrO (279.26): calcd. C 55.91, H 9.75; found C 56.03, H 9.90. M.p. 56 °C (ref.<sup>[14]</sup> 56–57 °C).

**12-Bromododecanol** (9b): Prepared from dodecanediol (5 g, 24.7 mmol) and hydrobromic acid (3.1 mL 48% aqueous solution, 27.5 mmol) in the same way as 9a. The crude product was purified by kugelrohr distillation (150 °C/0.4 mbar) and yielded 4.0 g (60%) of 9b as a colourless oil.  $^{1}$ H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.22–1.48 (m, 18 H, 9×CH<sub>2</sub>), 1.80 (pseudo quint, J = 7.0 Hz, 2 H, CH<sub>2</sub>CH<sub>2</sub>OH), 3.36 (t, J = 6.4 Hz, 2 H, CH<sub>2</sub>Br), 3.52 (t, J = 6.6 Hz, 2 H, CH<sub>2</sub>OH), 4.31 (t, J = 6.6 Hz, 1 H, OH; interchangeable with D<sub>2</sub>O) ppm. IR (NaCl):  $\tilde{v}$  = 3298 cm<sup>-1</sup> s, 2918 s, 2849 s, 1462 s, 1334 w, 1206 m, 1072 s, 1030 s, 938 m, 729 m, 651 s. C<sub>12</sub>H<sub>25</sub>BrO (265.23): calcd. C 54.34, H 9.50; found C 54.48, H 9.60.

**10-Bromodecanol (9d):** Prepared from decanediol (5 g, 28.7 mmol) and hydrobromic acid (3.6 mL 48% aqueous solution, 31.6 mmol) in the same way as **9a**. The crude product was purified by kugelrohr distillation (120 °C/0.5 mbar) and yielded 6.05 g (89%) of **9d** as a colourless oil. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.26–1.48 (m, 14 H, 7 × CH<sub>2</sub>), 1.80 (pseudo quint, J = 6.9 Hz, 2 H, CH<sub>2</sub>CH<sub>2</sub>OH), 3.38 (t, J = 6.4 Hz, 2 H, CH<sub>2</sub>Br), 3.52 (t, J = 6.7 Hz, 2 H, CH<sub>2</sub>OH), 3.68 (br. s, 1 H, OH; interchangeable with D<sub>2</sub>O) ppm. IR (NaCl):  $\bar{\nu}$  = 3332 cm<sup>-1</sup> s, 2926 s, 2853 s, 1464 s, 1370 w, 1256 m, 1056 s, 722 w, 644 w. C<sub>10</sub>H<sub>21</sub>BrO (237.18): calcd. C 50.64, H 8.92; found C 50.39, H 8.85.

**9-Bromononanol (9e):** Prepared from nonanediol (5 g, 31.2 mmol) and hydrobromic acid (3.9 mL 48% aqueous solution, 34.3 mmol) in the same way as **9a**. Kugelrohr distillation (120 °C/0.66 mbar) yielded 5.98 g (86%) of **9e** as a colourless oil. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.23–1.45 (m, 12 H, 6×CH<sub>2</sub>), 1.80 (quint, J = 6.9 Hz, 2 H, CH<sub>2</sub>CH<sub>2</sub>OH), 3.39 (t, J = 6.4 Hz, 2 H, CH<sub>2</sub>Br), 3.53 (t, J = 6.7 Hz, 2 H, CH<sub>2</sub>OH), 3.8 (br. s, 1 H, OH; interchangeable with D<sub>2</sub>O) ppm. IR (KBr):  $\tilde{v}$  = 3364 cm<sup>-1</sup> m, 2929 s, 2854 s, 1465 m, 1348 w, 1253 w, 1216 w, 1056 m, 1013 w, 726 w, 646 m, 562 w. C<sub>9</sub>H<sub>19</sub>BrO (223.15): calcd. C 48.44, H 8.58; found C 48.59, H 8.67.

**2-(13-Bromotridecyloxy)tetrahydro-2***H***-pyran (10a):** DHP (0.74 mL, 8.05 mmol) and PPTS (13.5 mg, 0.05 mmol) were added to a solution of **9a** (1.5 g, 5.37 mmol) in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> and the mixture was stirred at room temperature for 16 h. The solution was then washed with 2 m Na<sub>2</sub>CO<sub>3</sub> (2×15 mL) and the organic layer was dried with MgSO<sub>4</sub>. Evaporation of the solvent yielded crude **10a** as a yellow oil. Flash chromatography (19:1 hexane/ethyl acetate) yielded 1.52 g (78%) of **10a** as a colourless oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.15–1.85 (m, 28 H, 14×CH<sub>2</sub>), 3.26–3.48 (m, 4 H, CH<sub>2</sub>Br, CHH'O), 3.61–3.86 (m, 2 H, CHH'–O), 4.50 (pseudo t, *J* = 3.5 Hz, 1 H, C*H*) ppm. IR (NaCl):  $\tilde{v}$  = 2926 cm<sup>-1</sup> s, 2853 s, 1465 m, 1352 m, 1260 w, 1200 m, 1136 m, 1120 m, 1078 m, 1034 s, 987 w, 905 w, 869 m, 815 w, 722 w, 645 w. C<sub>18</sub>H<sub>35</sub>BrO<sub>2</sub> (363.37): calcd. C 59.50, H 9.71; found C 59.67, H 9.88.

**2-(12-Bromododecyloxy)tetrahydro-2***H***-pyran (10b):** Prepared from **9b** (3 g, 11.3 mmol), DHP (1.54 mL, 17.0 mmol) and PPTS (0.03 g, 0.11 mmol) in the same way as **10a**. Flash chromatography (19:1 hexane/ethyl acetate) yielded 3.5 g (90%) of **10b** as a colourless oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.25–1.96 (m, 26 H, 13×CH<sub>2</sub>), 3.36–3.57 (m, 4 H, CH<sub>2</sub>Br, CHH'O), 3.70–3.96 (m, 2 H, CHH'O), 4.60 (pseudo t, J = 3.5 Hz, 1 H, CH) ppm. IR (NaCl):  $\tilde{v}$  = 2926 cm<sup>-1</sup> s, 2854 s, 1465 m, 1352 m, 1260 w, 1200 m, 1120 m, 1078 m, 1034 s, 986 w, 905 w, 869 w, 815 w, 722 w. C<sub>17</sub>H<sub>33</sub>BrO<sub>2</sub> (349.35): calcd. C 58.45, H 9.52; found C 58.60, H 9.43.

**2-(11-Bromoundecyloxy)tetrahydro-2***H***-pyran (10c):** Prepared from bromoundecanol (2 g, 7.96 mmol), DHP (1.1 mL, 11.9 mmol) and PPTS (20 mg, 0.08 mmol) in the same way as **10a**. Flash chromatography (19:1 hexane/ethyl acetate) yielded 2.5 g (93%) of **10c** as a colourless oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.25–1.91 (m, 24 H, 12×CH<sub>2</sub>), 3.33–3.56 (m, 4 H, CH<sub>2</sub>Br, CHH'O), 3.68–3.93 (m, 2 H, CHH'O), 4.57 (pseudo t, J = 3.5 Hz, 1 H, CH) ppm. IR (NaCl):  $\tilde{v}$  = 2925 cm<sup>-1</sup> s, 2853 s, 1465 m, 1440 m, 1352 m, 1260 w, 1200 m, 1120 s, 1078 s, 1033 s, 988 w, 905 m, 869 m, 815 m, 722 w, 646 w. C<sub>16</sub>H<sub>31</sub>BrO<sub>2</sub> (335.32): calcd. C 57.31, H 9.32; found C 57.50, H 9.13.

**2-(10-Bromodecyloxy)tetrahydro-2***H***-pyran (10d):** Prepared from **9d** (4 g, 17 mmol), DHP (2.32 mL, 25.5 mmol) and PPTS (0.04 g, 0.17 mmol) in the same way as **10a**. Flash chromatography (19:1 hexane/ethyl acetate;  $R_{\rm f}=0.28$ ) yielded 4.4 g (80%) of **10d** as a colourless oil. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta=1.25-1.86$  (m, 22 H,  $11\times$  CH<sub>2</sub>), 3.27–3.79 (m, 6 H, CH<sub>2</sub>Br,  $2\times$  CH<sub>2</sub>O), 4.51–4.56 (m, 1 H, C*H*) ppm. IR (NaCl):  $\tilde{\rm v}=2927~{\rm cm}^{-1}$  s, 2854 s, 1465 m, 1440 m, 1352 m, 1260 m, 1200 m, 1120 m, 1078 m, 1033 s, 988 m, 905 w, 869 w, 815 w, 722 w. C<sub>15</sub>H<sub>29</sub>BrO<sub>2</sub> (321.29): calcd. C 56.07, H 9.10; found C 56.18, H 9.18.

**2-(9-Bromononyloxy)tetrahydro-2***H***-pyran (10e):** Prepared from **9e** (4 g, 17.89 mmol) DHP (2.46 mL, 26.9 mmol) and PPTS (0.045 g, 0.18 mmol) in the same way as **10a**. Flash chromatography (19:1 hexane/ethyl acetate;  $R_{\rm f}=0.3$ ) yielded 4.42 g (80%) of **10e** as a colourless oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta=1.28-1.93$  (m, 20 H,  $10\times$  CH<sub>2</sub>), 3.36–3.94 (m, 6 H, CH<sub>2</sub>Br,  $2\times$  CH<sub>2</sub>O), 4.58–4.61 (m, 1 H, C*H*) ppm. IR (NaCl):  $\tilde{v}=2932$  cm<sup>-1</sup> s, 2855 s, 1718 w, 1458 m, 1352 m, 1260 m, 1200 m, 1120 m, 1078 m, 1033 s, 989 w, 905 w, 869 w, 815 w. C<sub>14</sub>H<sub>27</sub>BrO<sub>2</sub> (307.27): calcd. C 54.72, H 8.86; found C 54.82, H 8.88.

3-[14-(Tetrahydro-2*H*-pyran-2-yloxy)tetradecyl|pyridine (11a): A solution of 3-picoline (14.3 mmol, 1.4 mL) in 20 mL of dry THF at -80 °C under argon was treated with LDA (7.15 mL of a 2 m solution in 10 mL of THF) over a period of 10 min. After 30 min of stirring a solution of 10a (7.15 mmol, 2.6 g) in 20 mL of dry THF was added within 10 min. Stirring was continued for 14 h during which time the mixture was warmed up to room temperature. After addition of 20 mL of sat. NH<sub>4</sub>Cl solution and 20 mL of water the layers were separated and the aqueous layer was extracted twice with 25 mL of ethyl acetate. The combined organic layers were dried with MgSO<sub>4</sub>. After evaporation of the solvent flash chromatography of the crude product (2:1 hexane/ethyl acetate;  $R_f = 0.36$ ) afforded 1.45 g (54%) of pure **11a** as a yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta = 1.22-1.39$  (m, 22 H,  $11 \times \text{CH}_2$ ), 1.47 - 1.928 H, py- $CH_2CH_2$ ,  $CH_2CH_2O$ (m,  $OCHCH_2CH_2CH_2$ ), 2.59 (t, J = 7.7 Hz, 2 H, py- $CH_2$ ), 3.34–3.54 (m, 2 H, OCHH'), 3.67-3.92 (m, 2 H, OCHH'), 4.57 (pseudo t, J = 3.5 Hz, OCHO), 7.19 (dd, J = 7.7, J = 4.8 Hz, 1 H, pyH5), 7.46 (dt, J = 7.8, J = 1.7 Hz, 1 H, pyH4), 8.40-8.46 (m, 2 H, pyH2 and)py*H6*) ppm. IR (NaCl):  $\tilde{v} = 2925 \text{ cm}^{-1} \text{ s}, 2853 \text{ s}, 1576 \text{ w}, 1466 \text{ m},$ 1421 m, 1352 w, 1322 w, 1260 w, 1200 w, 1121 m, 1078 m, 1033 s, 989 w, 905 w, 869 w, 814 w, 714 m. C<sub>24</sub>H<sub>41</sub>NO<sub>2</sub> (375.59): calcd. C 76.75, H 11.00, N 3.73; found C 76.71, H 11.21, N 3.84.

**3-[13-(Tetrahydro-2***H***-pyran-2-yloxy)tridecyl]pyridine (11b):** Prepared from 3-picoline (17.3 mmol, 1.68 mL), LDA (8.6 mL of a 2 m solution) and **10b** (5.75 mmol, 2 g) in the same way as **11a**. Flash chromatography of the crude product (2:1 hexane/ethyl acetate;  $R_{\rm f}$  = 0.26) afforded 1.45 g (70%) of pure **11b** as a yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.24–1.40 (m, 20 H, 10×CH<sub>2</sub>), 1.48–1.89 (m, 8 H, py-CH<sub>2</sub>CH<sub>2</sub>, CH<sub>2</sub>CH<sub>2</sub>O and OCHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 2.62 (t, J = 7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.35–3.56 (m, 2 H, OCHH'), 3.70–3.94

(m, 2 H, OCHH'), 4.60 (pseudo t, J = 3.5 Hz, OCHO), 7.21 (dd, J = 7.8, J = 4.7 Hz, 1 H, pyH5) 7.51 (dt, J = 7.8, J = 2.0 Hz, 1 H, pyH4), 8.42–8.48 (m, 2 H, pyH2 and pyH6) ppm. IR (NaCl):  $\tilde{v} = 2922$  cm $^{-1}$  s, 2853 s, 1574 m, 1465 m, 1422 m, 1352 m, 1260 w, 1200 m, 1135 s, 1078 s, 1033 s, 987 w, 905 w, 869 w, 815 w, 714 s. C $_{23}$ H $_{39}$ NO $_{2}$  (361.56): calcd. C 76.4, H 10.87, N 3.87; found C 76.41, H 10.93, N 3.89.

**3-[12-(Tetrahydro-2***H***-pyran-2-yloxy)dodecyl]pyridine (11c):** Prepared from 3-picoline (14.9 mmol, 1.5 mL), LDA (7.45 mL 2 M solution) and **10c** (2.5 g, 7.45 mmol) in the same way as **11a**. Flash chromatography of the crude product (2:1 hexane/ethyl acetate) afforded 1.44 g (59%) of pure **11c** as a yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.21–1.40 (m, 18 H, 9×CH<sub>2</sub>), 1.45–1.90 (m, 8 H, py-CH<sub>2</sub>CH<sub>2</sub>, CH<sub>2</sub>CH<sub>2</sub>O and OCHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 2.60 (t, J = 7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.33–3.54 (m, 2 H, OCHH'), 3.68–3.92 (m, 2 H, OCHH'), 4.57 (pseudo t, J = 3.5 Hz, OCHO), 7.19 (dd, J = 7.8, J = 4.8 Hz, 1 H, py*H*5), 7.48 (dt, J = 7.8, J = 2.2 Hz, 1 H, py*H*4), 8.40–8.45 (m, 2 H, py*H*2 and py*H*6) ppm. IR (NaCl):  $\bar{v}$  = 2926 cm<sup>-1</sup> s, 2854 s, 1576 w, 1466 w, 1421 w, 1352 w, 1200 w, 1136 m, 1120 m, 1078 m, 1034 m, 869 w, 815 w, 714 m. C<sub>22</sub>H<sub>37</sub>NO<sub>2</sub> (347.53): calcd. C 76.03, H 10.73, N 4.03; found C 75.66, H 10.96, N 4.40.

**3-[11-(Tetrahydro-2***H***-pyran-2-yloxy)undecyl[pyridine (11d):** Prepared from 3-picoline (12.4 mmol, 1.2 mL), LDA (6.2 mL of a 2 M solution) and **10d** (6.2 mmol, 2 g) in the same way as **11a**. Flash chromatography of the crude product (2:1 hexane/ethyl acetate;  $R_{\rm f}=0.35$ ) afforded 1.45 g (70%) of pure **11d** as a yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta=1.32$  (br. s, 16 H, 8×CH<sub>2</sub>), 1.48–1.89 (m, 8 H, py-CH<sub>2</sub>CH<sub>2</sub>, CH<sub>2</sub>CH<sub>2</sub>O and OCHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 2.61 (t, J=7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.35–3.56 (m, 2 H, OCHH'), 3.70–3.94 (m, 2 H, OCHH'), 4.59 (pseudo t, J=3.5 Hz, OCHO), 7.21 (dd, J=7.6 Hz, 1 H, pyH5) 7.51 (dt, J=7.8, J=1.9 Hz, 1 H, pyH4), 8.43–8.47 (m, 2 H, pyH2 and pyH6) ppm. IR (NaCl):  $\tilde{v}=2926$  cm<sup>-1</sup> s, 2853 s, 1574 m, 1465 m, 1421 m, 1352 m, 1260 w, 1200 w, 1120 m, 1078 m, 1033 s, 986 w, 905 m, 869 m, 815 m, 713 s. C<sub>21</sub>H<sub>35</sub>NO<sub>2</sub> (333.51): calcd. C 75.63, H 10.58, N 4.20; found C 75.78, H 10.67, N 4.37.

**3-[10-(Tetrahydro-2***H***-pyran-2-yloxy)decyl|pyridine (11e):** Prepared from 3-picoline (2.85 mL, 29.3 mmol), LDA (14.7 mL of a 2 M solution) and **10e** (3 g, 9.76 mmol) as described above for **11a**. Flash chromatography of the crude product (2:1 hexane/ethyl acetate;  $R_{\rm f}=0.35$ ) afforded 1.82 g (57%) of pure **11e** as a yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta=1.28$  (br. s, 14 H,  $7\times$  CH<sub>2</sub>), 1.47–1.87 (m, 8 H, py-CH<sub>2</sub>CH<sub>2</sub>, CH<sub>2</sub>CH<sub>2</sub>O and OCHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>), 2.59 (t, J=7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.32–3.53 (m, 2 H, OCHH'), 3.67–3.91 (m, 2 H, OCHH'), 4.56 (pseudo t, J=3.5 Hz, OCHO), 7.19 (dd, J=7.8, J=4.8 Hz, 1 H, pyH5), 7.47 (dt, J=7.8, J=1.9 Hz, 1 H, pyH4), 8.40–8.43 (m, 2 H, pyH2 and pyH6) ppm. IR (NaCl):  $\tilde{v}=2927$  cm<sup>-1</sup> s, 2854 s, 1575 w, 1465 m, 1422 m, 1352 m, 1260 w, 1200 m, 1120 m, 1078 m, 1033 s, 905 w, 869 w, 814 w, 714 m. C<sub>20</sub>H<sub>33</sub>NO<sub>2</sub> (319.48): calcd. C 75.19, H 10.41, N 4.38; found C 74.93, H 10.28, N 4.58.

**14-(Pyridin-3-yl)tetradecanol (12a):** Compound **11a** (1.29 g, 3.44 mmol) was dissolved in 50 mL of methanol and 6.9 mL of 1 m HCl was added. After stirring for 12 h at room temperature the solution was concentrated to 1/4 of its volume. The pH was brought to 10 by addition of 2 m NaOH and the solution was extracted three times with 10 mL of ethyl acetate. The combined organic layers were dried with MgSO<sub>4</sub> and the solvent removed to yield 0.96 g (96%) of **12a**. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.18–1.46 (m, 2 H, 11×CH<sub>2</sub>), 1.48–1.64 (m, 2 H,  $CH_2$ CH<sub>2</sub>OH), 2.57 (t, D = 7.6 Hz, 2 H, py- $CH_2$ ), 3.37 (dt and t, D = 6.2, D = 5.1 Hz, 2 H,

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C $H_2$ OH), 4.31 (t, J = 5.1 Hz, 1 H, OH; interchangeable with D<sub>2</sub>O), 7.28 (dd, J = 7.8, J = 4.8 Hz, 1 H, pyH5), 7.60 (dt, J = 7.7, J = 2.2 Hz, 1 H, pyH4), 8.38 (dd, J = 4.8, J = 1.7 Hz, 1 H, pyH6), 8.40 (d, J = 1.7 Hz, 1 H, pyH2) ppm. IR (KBr):  $\tilde{v}$  = 3334 cm $^{-1}$  s, 2917 s, 2849 s, 1578 m, 1472 s, 1423 s, 1338 m, 1228 w, 1186 w, 1114 w, 1070 m, 1029 w, 1011 w, 964 w, 928 m, 888 w, 831 w, 802 m, 716 s, 640 m. C<sub>19</sub>H<sub>33</sub>NO (291.47): calcd. C 78.29, H 11.41, N 4.81; found C 78.03, H 11.39, N 4.77. M.p. 61 °C (ref. [9] 61–62 °C).

**13-(Pyridin-3-yl)tridecanol (12b):** Prepared from **11b** (1.45 g, 4.0 mmol) and 8.0 mL of 1 m HCl as described above for **12a**. Recrystallisation from 1:1 hexane/ethyl acetate yielded 0.84 g (76%) of **12b**. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta = 1.14$ –1.33 (m, 18 H, 9×CH<sub>2</sub>), 1.33–1.48 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>py), 1.48–1.64 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>OH), 2.58 (t, J = 7.5 Hz, 2 H, py-CH<sub>2</sub>), 3.36–3.42 (m, 2 H, CH<sub>2</sub>OH), 4.32 (t, J = 5.1 Hz, 2 H, OH; interchangeable with D<sub>2</sub>O), 7.29 (dd, J = 7.8, J = 4.7 Hz, 1 H, pyH5), 7.61 (dt, J = 7.8, J = 1.9 Hz, 1 H, pyH4), 8.39 (dd, J = 4.7, J = 1.6 Hz, 1 H, pyH6), 8.42 (d, J = 1.6 Hz, 1 H, pyH2) ppm. IR (KBr):  $\tilde{v} = 3241$  cm<sup>-1</sup> m, 2923 s, 2848 s, 1574 m, 1470 m, 1420 m, 1378 w, 1186 w, 1078 s, 1025 m, 908 w, 802 m, 707 m, 634 w. C<sub>18</sub>H<sub>31</sub>NO (277.44): calcd. C 77.92, H 11.26, N 5.05; found C 77. 66, H 11.33, N 4.86. M.p. 48 °C (ref. [9] 48–50 °C).

**12-(Pyridin-3-yl)dodecanol (12c):** Prepared from **11c** (2.16 g, 6.63 mmol) and 13.3 mL of 1 M HCl as described above for **12a**. Recrystallisation from 1:1 hexane/ethyl acetate yielded 1.36 g (78%) of **12c**. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.21–1.39 (m, 16 H, 8×CH<sub>2</sub>), 1.50–1.68 (m, 4 H, CH<sub>2</sub>CH<sub>2</sub>py, CH<sub>2</sub>CH<sub>2</sub>OH), 2.59 (t, J = 7.6 Hz, 2 H, py-CH<sub>2</sub>), 3.36 (t, J = 6.6 Hz, 2 H, CH<sub>2</sub>OH), 7.19 (dd, J = 4.8, J = 7.8 Hz, 1 H, pyH5), 7.48 (dt, J = 7.8, J = 1.8 Hz, 1 H, pyH4), 8.40–8.45 (m, 2 H, pyH2, pyH6) ppm. IR (KBr):  $\tilde{v}$  = 3334 cm<sup>-1</sup> s, 2918 s, 2850 s, 1578 m, 1472 m, 1423 m, 1335 w, 1186 w, 1113 w, 1071 m, 1027 w, 990 w, 947 w, 903 w, 830 w, 803 m, 715 s, 639 m. C<sub>17</sub>H<sub>29</sub>NO (263.42): calcd. C 77.51, H 11.10, N 5.32; found C 77.54, H 11.27, N 5.46. M.p. 48–49 °C (ref.<sup>[15]</sup> 48–50 °C).

**11-(Pyridin-3-yl)undecanol (12d):** Prepared from **11d** (1.45 g, 4.35 mmol) and 8.7 mL of 1 m HCl as described above for **12a**. Drying under high vacuum yielded 0.87 g (80%) of **12d**. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.14–1.46 (m, 16 H, 8×CH<sub>2</sub>), 1.50–1.63 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>OH), 2.59 (t, J = 7.6 Hz, 2 H, py-CH<sub>2</sub>), 3.34–3.41 (m, 2 H, CH<sub>2</sub>OH), 4.33 (t, J = 5.1 Hz, 1 H, OH; interchangeable with D<sub>2</sub>O), 7.30 (dd, J = 7.8, J = 4.8 Hz, 1 H, pyH5), 7.62 (dt, J = 7.8, J = 1.9 Hz, 1 H, pyH4), 8.39 (dd, J = 4.8, J = 1.6 Hz, 1 H, pyH6), 8.43 (d, J = 2 Hz, 1 H, pyH2) ppm. IR (KBr):  $\tilde{v}$  = 3245 cm<sup>-1</sup> m, 2926 s, 2852 s, 1576 m, 1466 m, 1420 s, 1376 w, 1188 m, 1078 s, 1019 m, 934 w, 801 m, 709 s, 635 m. C<sub>16</sub>H<sub>27</sub>NO (249.39): calcd. C 77.06, H 10.91, N 5.62; found C 76.88, H 10.91, N 5.85.

**10-(Pyridin-3-yl)decanol** (**12e):** Prepared from **11e** (1.8 g, 5.63 mmol) and HCl (11.2 mL 1 m solution) in the same way as **12a**. Flash chromatography (19:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol;  $R_f = 0.37$ ) of the crude product gave 1.23 g (93%) of **12e** as a yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta = 1.22-1.40$  (m, 12 H,  $6 \times$  CH<sub>2</sub>), 1.49–1.67 (m, 4 H, CH<sub>2</sub>CH<sub>2</sub>OH and py-CH<sub>2</sub>CH<sub>2</sub>), 1.90 (br. s, 1 H, OH), 2.59 (t, J = 7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.63 (t, J = 6.6 Hz, 2 H, CH<sub>2</sub>OH), 7.19 (dd, J = 7.8, J = 4.8 Hz, 1 H, pyH5), 7.48 (dt, J = 7.8, J = 1.9 Hz, 1 H, pyH4), 8.39–8.44 (m, 2 H, pyH6 and pyH2) ppm. IR (NaCl):  $\tilde{v} = 3331$  cm<sup>-1</sup> s, 2925 s, 2853 s, 1576 m, 1465 s, 1423 s, 1369 w, 1190 w, 1058 s, 1028 s, 924 w, 794 m, 713 s. C<sub>15</sub>H<sub>25</sub>NO (235.37): calcd. C 76.55, H 10.71, N 5.95; found C 76.62, H 10.89, N 6.16.

**3-(13-Chlorotridecyl)pyridine (13b):** A solution of **12b** (1.0 g, 3.6 mmol) in 5 mL of dioxane was added to SOCl<sub>2</sub> (0.32 mL,

4.3 mmol) at 0 °C within 10 min. After stirring at room temperature for 1 h, 10 mL of ethanol was added and the mixture was heated to reflux for 10 min. The solvent was evaporated and the residue dissolved in 10 mL of CH<sub>2</sub>Cl<sub>2</sub>. The solution was washed three times with 10 mL of 2 M Na<sub>2</sub>CO<sub>3</sub>, the organic layer dried with MgSO<sub>4</sub> and the solvent removed. Flash chromatography (3:1 hexane/ethyl acetate;  $R_f = 0.24$ ) of crude 13b gave 753 mg (71%) of the pure compound as a slightly yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta = 1.25-1.50$  (m, 18 H,  $9 \times \text{CH}_2$ ), 1.55-1.85 (m, 4 H, py- $CH_2CH_2$  and  $CH_2CH_2Cl$ ), 2.63 (t, J = 7.7 Hz, 2 H, py- $CH_2$ ), 3.55 (t, J = 6.7 Hz, 2 H,  $CH_2Cl$ ), 7.23 (dd, J = 7.8, J = 4.7 Hz, 1 H, pyH5), 7.51 (dt, J = 7.8, J = 1.9 Hz, 1 H, pyH4), 8.43–8.49 (m, 2 H, pyH6 and pyH2) ppm. IR (NaCl):  $\tilde{v} = 2926 \text{ cm}^{-1} \text{ s}$ , 2853 s, 2365 w, 1574 m, 1465 m, 1421 m, 1308 w, 1189 w, 1127 w, 1026 m, 793 w, 714 m. C<sub>18</sub>H<sub>30</sub>ClN (295.89): calcd. C 73.07, H 10.22, N 4.73; found C 73.24, H 10.32, N 4.70.

**3-(11-Chloroundecyl)pyridine** (13d): Prepared from 12d (0.8 g, 3.21 mmol) and SOCl<sub>2</sub> (0.28 mL, 3.85 mmol) as described above for 13b. Flash chromatography (2:1 hexane/ethyl acetate;  $R_{\rm f}=0.33$ ) of crude 13d gave 645 mg (75%) of the pure compound as a slightly brown oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta=1.23-1.47$  (m, 14 H,  $7 \times$  CH<sub>2</sub>), 1.55–1.67 (m, 2 H, py-CH<sub>2</sub>CH<sub>2</sub>), 1.76 (quint, J=7.1 Hz, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl), 2.59 (t, J=7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.53 (t, J=6.7 Hz, 2 H, CH<sub>2</sub>Cl), 7.20 (dd, J=7.8, J=4.8 Hz, 1 H, pyH5), 7.49 (dt, J=7.8, J=1.9 Hz, 1 H, pyH4), 8.41–8.45 (m, 2 H, pyH6 and pyH2) ppm. IR (NaCl):  $\tilde{v}=2926$  cm<sup>-1</sup> s, 2853 s, 1710 w, 1574 s, 1465 s, 1422 s, 1308 m, 1189 w, 1128 w, 1026 s, 793 m, 714 s, 651 m. C<sub>16</sub>H<sub>26</sub>ClN (267.84): calcd. C 71.75, H 9.78, N 5.23; found C 71.83, H 9.86, N 5.35

**3-(10-Chlorodecyl)pyridine (13e):** Prepared from **12e** (1.56 g, 6.62 mmol) and SOCl<sub>2</sub> (0.72 mL, 9.93 mmol) as described above for **13b**. Flash chromatography (2:1 hexane/ethyl acetate;  $R_{\rm f} = 0.45$ ) of crude **12e** gave 1.32 g (79%) of pure **13e** as an orange oil.  $^{1}{\rm H}$  NMR (250 MHz, CDCl<sub>3</sub>):  $\delta = 1.18-1.41$  (m, 12 H,  $6 \times {\rm CH_2}$ ), 1.46–1.61 (m, 2 H, py-CH<sub>2</sub>CH<sub>2</sub>), 1.63–1.79 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl), 2.53 (t, J = 7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.46 (t, J = 6.7 Hz, 2 H, CH<sub>2</sub>Cl), 7.13 (dd, J = 4.8, J = 7.8 Hz, 1 H, pyH5), 7.41 (dt, J = 1.7, J = 7.8 Hz, 1 H, pyH4), 8.34–8.39 (m, 2 H, pyH6, pyH2) ppm. IR (NaCl):  $\tilde{v} = 2926$  cm<sup>-1</sup> s, 2853 s, 2216 m, 1734 w, 1574 s, 1464 s, 1422 s, 1308 m, 1189 m, 1128 w, 1107 w, 1026 s, 910 s, 793 m, 732 s, 713 s, 649 m. C<sub>15</sub>H<sub>24</sub>ClN (253.81): calcd. C 70.98, H 9.53, N 5.52; found C 70.89, H 9.54, N 5.63

**3-(13-Chlorotridecyl)pyridine 1-Oxide (14b):** *m*-Chloroperbenzoic acid (710 mg, 2.78 mmol, 77% active) was added to a solution of **13b** (750 mg, 2.53 mmol) in 20 mL of CH<sub>2</sub>Cl<sub>2</sub> at 0 °C and the mixture was stirred for 1 h. After concentrating the solution to approximately 2 mL it was column-filtered through basic Al<sub>2</sub>O<sub>3</sub> (40 g, 19:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol). Evaporation of the solvent gave 739 mg (93%) of **14b** as a white solid. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.22–1.50 (m, 18 H, 9×CH<sub>2</sub>), 1.57–1.68 (m, 2 H, py-CH<sub>2</sub>CH<sub>2</sub>), 1.79 (quint, J = 7.1 Hz, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl), 2.59 (t, J = 7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.55 (t, J = 6.7 Hz, 2 H, CH<sub>2</sub>Cl), 7.10–7.25 (m, 2 H, pyH5 and pyH4), 8.08–8.12 (m, 2 H, pyH6 and pyH2) ppm. IR (KBr):  $\tilde{v}$  = 3084 cm<sup>-1</sup> m, 2918 s, 2850 s, 1599 m, 1560 m, 1472 s, 1420 s, 1329 w, 1262 s, 1207 w, 1151 s, 1014 s, 939 s, 856 m, 793 m, 718 s, 678 s. C<sub>18</sub>H<sub>30</sub>ClNO (311.89): calcd. C 69.32, H 9.70, N 4.39; found C 69.31, H 9.77, N 4.39. M.p. 63 °C (ref. [9] 62–63 °C).

**3-(11-Chloroundecyl)pyridine** *N***-Oxide (14d):** Prepared from **13d** (471 mg, 1.76 mmol) and *m*-chloroperbenzoic acid (434 mg, 1.94 mmol, 77% active) as described above for **14b**. Evaporation of the solvent gave 404 mg (81%) of **14d** as a white solid. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta = 1.23-1.47$  (m, 14 H,  $7 \times$  CH<sub>2</sub>), 1.59–1.67

(m, 2 H, py-CH<sub>2</sub>CH<sub>2</sub>), 1.79 (quint, J=7.1 Hz, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl), 2.60 (t, J=7.7 Hz, 2 H, py-CH<sub>2</sub>), 3.56 (t, J=6.7 Hz, 2 H, CH<sub>2</sub>Cl), 7.10–7.25 (m, 2 H, pyH5 and pyH4), 8.05–8.13 (m, 2 H, pyH6 and pyH2) ppm. IR (KBr):  $\tilde{v}=3084$  cm<sup>-1</sup> m, 2982 s, 2851 s, 1599 m, 1560 m, 1472 s, 1420 s, 1309 m, 1268 s, 1213 m, 1152 s, 1010 s, 940 s, 862 m, 794 m, 718 s, 677 s. C<sub>16</sub>H<sub>26</sub>ClNO (283.84): calcd. C 67.70, H 9.23, N 4.93; found C 67.70, H 8.97, N 4.77. M.p. 49–50 °C.

**3-(10-Chlorodecyl)pyridine** *N*-Oxide **(14e):** Prepared from **13e** (1.46 g, 5 mmol) and *m*-chloroperbenzoic acid (1.29 g, 7.5 mmol, 77% active) as described above for **14b**. Evaporation of the solvent gave 1.14 g (83%) of **14e** as a slightly yellow oil. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.23–1.48 (m, 12 H, 6×CH<sub>2</sub>), 1.51–1.68 (m, 2 H, py-CH<sub>2</sub>CH<sub>2</sub>), 1.70–1.83 (m, 2 H, CH<sub>2</sub>CH<sub>2</sub>Cl), 2.56 (t, *J* = 7.6 Hz, 2 H, py-CH<sub>2</sub>), 3.52 (t, *J* = 6.7 Hz, 2 H, CH<sub>2</sub>Cl), 7.05–7.21 (m, 2 H, pyH5 and pyH4), 8.03–8.10 (m, 2 H, pyH6 and pyH2) ppm. IR (NaCl):  $\tilde{v}$  = 3064 cm<sup>-1</sup> w, 2227 s, 2854 s, 1602 m, 1506 m, 1480 w, 1436 s, 1273 s, 1159 s, 1014 s, 966 w, 862 w, 793 w, 761 w, 722 w, 681 m. C<sub>15</sub>H<sub>24</sub>ClNO (269.81): calcd. C 66.77, H 8.97, N 5.19 calcd. **13e**·0.25 H<sub>2</sub>O: calcd. C 65.68, H 9.00, N 5.11; found C 65.69, H 9.15, N 5.13.

3-(12-Hydroxydodecyl)-1-[10-(1-oxidopyridin-3-yl)decyl|pyridinium **Iodide (15f):** Compound 14e (830 mg, 3.07 mmol) was added to a solution of 12c (0.81 g, 3.07 mmol) and NaI (552 mg, 3.7 mmol) in 30 mL of butan-2-one and the mixture was refluxed for 40 h. The solvent was then removed and the residue was adsorbed on SiO<sub>2</sub>. Flash chromatography on SiO<sub>2</sub> (9:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol) yielded 1.27 g (66%) of **15f** as a waxy solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>] DMSO):  $\delta = 1.11-1.71$  (m, 34 H,  $17 \times \text{CH}_2$ ), 1.84-1.99 (m, 2 H,  $NCH_2CH_2$ ), 2.53 (t, J = 7.5 Hz, 2 H, 3- $CH_2$ ), 2.80 (t, J = 7.5 Hz, 2 H, 3'-C $H_2$ ), 3.32–3.40 (m, 2 H, C $H_2$ OH), 4.33 (t, J = 5.0 Hz, 1 H, OH; interchangeable with  $D_2O$ ), 4.56 (t, J = 7.3 Hz, 2 H,  $NCH_2$ ), 7.19 (d, J = 7.8 Hz, 1 H, H4), 7.34 (t, J = 7.2 Hz, 1 H, H5), 8.03–8.14 (m, 3 H, H2, H2' and H5'), 8.49 (d, J = 8.0 Hz, 1 H, H4'), 8.95 (d, J = 6.0 Hz, 1 H, H6'), 9.05 (s, 1 H, H2') ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 25.3, 25.4, 28.15, 28.20, 28.25, 28.55, 28.60, 28.65, 28.80, 28.85, 28.90 and 29.0 ( $12 \times CH_2$ ), 29.6 (3'-CH<sub>2</sub>CH<sub>2</sub>), 29.8 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.5 (CH<sub>2</sub>CH<sub>2</sub>N), 31.5 (3'-CH<sub>2</sub> + 3-CH<sub>2</sub>), 32.4 (CH<sub>2</sub>CH<sub>2</sub>OH), 60.6 (CH<sub>2</sub>OH, CH<sub>2</sub>N), 125.4 (C4), 125.9 (C5), 127.5 (C5'), 136.1 (C6), 138.0 (C2), 141.3 (C3'), 142.1 (C6'), 142.9 (C3), 143.8 (C2'), 145.0 (C4') ppm. IR (KBr): ṽ  $= 3422 \text{ cm}^{-1} \text{ s}, 3029 \text{ w}, 2924 \text{ s}, 2849 \text{ s}, 1630 \text{ w}, 1570 \text{ w}, 1500 \text{ m},$ 1459 m, 1353 w, 1259 m, 1166 s, 1107 w, 1060 m, 1023 w, 953 w, 810 m, 761 m, 725 w, 689 m. HRMS: calcd. for  $C_{32}H_{53}N_2O_2^+$ 497.4102; found 497.4108.

3-(10-Hydroxydecyl)-1-[13-(1-oxidopyridin-3-yl)tridecyl]pyridinium Iodide (15g): Prepared from 12e (0.67 g, 2.86 mmol), 14b (0.89 g, 2.86 mmol) and NaI (0.51 g, 3.4 mmol) as described above for 15f. Flash chromatography on SiO<sub>2</sub> (9:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol) yielded 1.1 g (60%) of 15g as orange solid.  $^{1}H$  NMR (250 MHz, [D<sub>6</sub>] DMSO):  $\delta = 1.11-1.45$  (m, 32 H,  $16 \times \text{CH}_2$ ), 1.46-1.71 (m, 4 H, 3'-CH<sub>2</sub>-CH<sub>2</sub>, 3-CH<sub>2</sub>-CH<sub>2</sub>), 1.83-1.98 (m, 2 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.53 (t, J = 7.4 Hz, 2 H, 3-C $H_2$ ), 2.79 (t, J = 7.6 Hz, 2 H, 3'-C $H_2$ ), 3.32–3.40 (m, 2 H,  $CH_2OH$ ), 4.30 (t, J = 5.1 Hz, 1 H, OH; interchangeable with  $D_2O$ ), 4.55 (t, J = 7.3 Hz, 2 H,  $NCH_2$ ), 7.19 (dt, J = 7.8, J =1.3 Hz, 1 H, H4), 7.34 (dd, J = 7.5, J = 6.5 Hz, 1 H, H5), 8.02– 8.12 (m, 3 H, H2, H2' and H5'), 8.48 (d, J = 8.2 Hz, 1 H, H4'), 8.93 (d, J = 6.0 Hz, 1 H, H6'), 9.03 (s, 1 H, H2') ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 25.2, 25.4, 28.1, 28.2, 28.3, 28.60, 28.65, 28.75, 28.80 and 28.9 (10 × CH<sub>2</sub>), 29.6 (3'-CH<sub>2</sub>CH<sub>2</sub>), 29.8 (3- $CH_2CH_2$ ), 30.5 ( $CH_2CH_2N$ ), 31.5 (3'- $CH_2$  + 3- $CH_2$ ), 32.4  $(CH_2CH_2OH)$ , 60.6  $(CH_2OH + CH_2N)$ , 125.4 (C4), 125.9 (C5), 127.5 (C5'), 136.5 (C6), 138.0 (C2), 141.3 (C3'), 142.1(C6'), 142.9

(C3), 143.8 (C2'), 145.5 (C4') ppm. IR (KBr):  $\bar{v}=3422~cm^{-1}~s$ , 2922 s, 2851 s, 1629 w, 1560 w, 1507 w, 1468 m, 1438 w, 1260 m, 1160 m, 1063 w, 1020 w, 962 w, 791 w, 760 w, 719 w, 678 m. HRMS: calcd. for  $C_{33}H_{55}N_2O_2^+$  511.4258; found 511.4262.

3-(14-Hydroxytetradecyl)-1-[10-(1-oxidopyridin-3-yl)decyl]pyridinium Iodide (15h): Prepared from 12a (0.9 g, 3.08 mmol), 14e (0.83 g, 3.08 mmol) and NaI (0.55 g, 3.7 mmol) as described above for 15f. Flash chromatography on SiO<sub>2</sub> (9:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol) yielded 1.57 g (78%) of **15h** as a yellow solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>] DMSO):  $\delta = 1.18-1.72$  (m, 38 H,  $19 \times \text{CH}_2$ ), 1.83-1.98 (m, 2 H,  $NCH_2CH_2$ ), 2.53 (t, J = 7.3 Hz, 2 H, 3- $CH_2$ ), 2.78 (t, J = 7.5 Hz, 2 H, 3'-C $H_2$ ), 3.31–3.40 (m, 2 H, C $H_2$ OH), 4.32 (t, J = 5.1 Hz, 1 H, OH; interchangeable with  $D_2O$ ), 4.56 (t, J = 7.3 Hz, 2 H,  $NCH_2$ ), 7.19 (d, J = 8.0 Hz, 1 H, H4), 7.32 (dd, J = 7.6, J = 6.6 Hz, 1 H, H5), 8.02-8.13 (m, 3 H, H2, H2' and H5'), 8.49 (d, J = 8.0 Hz, 1 H, H4'), 8.95 (d, J = 6.0 Hz, 1 H, H6'), 9.05 (s, 1 H, H2') ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 25.2, 25.4, 28.15, 28.20, 28.25, 28.55, 28.60, 28.62, 28.66, 28.80, 28.85, 28.94 and 29.0  $(13 \times CH_2)$ , 29.6  $(3'-CH_2CH_2)$ , 29.8  $(3-CH_2CH_2)$ , 30.5  $(CH_2CH_2N)$ , 31.5 (3'- $CH_2$  + 3- $CH_2$ ), 32.4 ( $CH_2CH_2OH$ ), 60.5  $(CH_2OH + CH_2N)$ , 125.4 (C4), 125.9 (C5), 127.5 (C5'), 136.2 (C6), 138.0 (C2), 141.3 (C3'), 142.1 (C6'), 142.9 (C3), 143.8 (C2'), 145.0 (C4') ppm. IR (KBr):  $\tilde{v} = 3384 \text{ cm}^{-1} \text{ s}$ , 3046 m, 2918 s, 2850 s, 1734 w, 1630 m, 1604 m, 1570 w, 1505 w, 1466 s, 1438 m, 1364 w, 1314 w, 1263 s, 1161 s, 1054 m, 1019 w, 964 w, 917 w, 804 m, 761 w, 721 m, 683 s. HRMS: calcd. for C<sub>34</sub>H<sub>57</sub>N<sub>2</sub>O<sub>2</sub><sup>+</sup> 525.4415; found 525.4381.

3-(12-Hydroxydodecyl)-1-[11-(1-oxidopyridin-3-yl)undecyl]pyridinium Iodide (15i): Prepared from 12c (0.74 g, 2.82 mmol), 14d (0.8 g, 2.82 mmol) and NaI (0.51 g, 3.4 mmol) as described above for 15f. Filtration of the precipitate yielded 1.53 g (85%) of 15i as an offwhite solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.15–1.71 (m, 36 H,  $18 \times \text{CH}_2$ ), 1.83-1.98 (m, 2 H,  $\text{NCH}_2\text{C}H_2$ ), 2.51 (t, J = 8.2 Hz, 2 H, 3-C $H_2$ ), 2.79 (t, J = 7.5 Hz, 2 H, 3'-C $H_2$ ), 3.34–3.40 (m, 2 H,  $CH_2OH$ ), 4.38 (t, J = 4.5 Hz, 1 H, OH; interchangeable with  $D_2O$ ), 4.58 (t, J = 7.3 Hz, 2 H, NC $H_2$ ), 7.19 (dt, J = 8.1, J = 1.3 Hz, 1 H, H4), 7.32 (dd, J = 6.5, J = 7.5 Hz, 1 H, H5), 8.02–8.12 (m, 3 H, H2, H2' and H5'), 8.49 (d, J = 8.1 Hz, 1 H, H4'), 9.00 (d, J =6.0 Hz, 1 H, H6'), 9.10 (s, 1 H, H2') ppm. <sup>13</sup>C NMR (62.5 MHz,  $[D_6]DMSO$ ):  $\delta = 25.2, 25.4, 28.2, 28.3, 28.55, 28.60, 28.66, 28.73,$ 28.77, 28.82, 28.85, 28.90 and 29.0 (14×CH<sub>2</sub>), 29.6 (3'-CH<sub>2</sub>CH<sub>2</sub>), 29.8 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.5 (CH<sub>2</sub>CH<sub>2</sub>N), 31.5 (3'-CH<sub>2</sub> + 3-CH<sub>2</sub>), 32.4  $(CH_2CH_2OH)$ , 60.5  $(CH_2OH + CH_2N)$ , 125.4 (C4), 125.9 (C5), 127.5 (C5'), 136.2 (C6), 138.1 (C2), 141.3 (C3'), 142.2 (C6'), 142.9 (C3), 143.9 (C2'), 145.0 (C4') ppm. IR (KBr):  $\tilde{v} = 3256 \text{ cm}^{-1} \text{ m}$ , 3067 w, 2917 s, 2848 s, 1630 w, 1603 w, 1560 w, 1492 w, 1468 m, 1431 m, 1330 w, 1269 m, 1215 m, 1156 s, 1079 m, 1016 w, 965 m, 906 w, 817 w, 801 w, 760 w, 720 w, 682 m. HRMS: calcd. for  $C_{33}H_{55}N_2O_2^+$  511.4258; found 511.4254.

**3-(14-Hydroxytetradecyl)-1-[11-(1-oxidopyridin-3-yl)undecyl]pyridinium Iodide (15j):** Prepared from **12a** (0.7 g, 2.4 mmol), **14d** (0.68 g, 2.4 mmol) and NaI (0.43 g, 2.88 mmol) as described above for **15f**. Flash chromatography on SiO<sub>2</sub> (9:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol) yielded 0.85 g (53%) of **15h** as a yellow solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.15–1.72 (m, 40 H, 20×CH<sub>2</sub>), 1.82–1.98 (m, 2 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.52 (t, J = 7.7 Hz, 2 H, 3-CH<sub>2</sub>), 2.78 (t, J = 7.5 Hz, 2 H, 3'-CH<sub>2</sub>), 3.32–3.40 (m, 2 H, CH<sub>2</sub>OH), 4.38 (t, J = 5.1 Hz, 1 H, OH; interchangeable with D<sub>2</sub>O), 4.55 (t, J = 7.3 Hz, 2 H, NCH<sub>2</sub>), 7.19 (d, J = 8.0 Hz, 1 H, H4), 7.32 (dd, J = 6.4, J = 7.6 Hz, 1 H, H5), 8.02–8.13 (m, 3 H, H2, H2' and H5'), 8.48 (d, J = 8.1 Hz, 1 H, H4'), 8.94 (d, J = 6.0 Hz, 1 H, H6'), 9.04 (s, 1 H, H2') ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 25.2, 25.4, 28.15, 28.23,

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28.27, 28.61, 28.66, 28.74, 28.77, 28.83, 28.86, 28.94 and 29.0 (13 × CH<sub>2</sub>), 29.6 (3'-CH<sub>2</sub>CH<sub>2</sub>), 29.8 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.5 (CH<sub>2</sub>CH<sub>2</sub>N), 31.5 (3'-CH<sub>2</sub> + 3-CH<sub>2</sub>), 32.4 (CH<sub>2</sub>CH<sub>2</sub>OH), 60.5 (CH<sub>2</sub>OH + CH<sub>2</sub>N), 125.4 (C4), 125.9 (C5), 127.5 (C5'), 136.2 (C6), 138.1 (C2), 141.3 (C3'), 142.1 (C6'), 142.9 (C3), 143.8 (C2'), 145.0 (C4') ppm. IR (KBr):  $\tilde{v}$  = 3405 cm<sup>-1</sup> s, 3048 w, 2918 s, 2850 s, 1629 w, 1604 w, 1560 w, 1507 w, 1466 m, 1438 w, 1351 w, 1263 s, 1161 s, 1054 m, 1019 w, 961 w, 800 w, 762 w, 721 w, 683 m. HRMS: calcd. for C<sub>35</sub>H<sub>59</sub>N<sub>2</sub>O<sub>2</sub>+539.4571; found 539.4519.

3-(10-Hydroxydecyl)-1-[11-(1-oxidopyridin-3-yl)undecyl|pyridinium **Iodide (15k):** Prepared from **12d** (0.66 g; 2.79 mmol) and **14e** (0.79 g; 2.79 mmol) as described for **15f**. Flash chromatography on SiO<sub>2</sub> (9:1 CH<sub>2</sub>Cl<sub>2</sub>/methanol) yielded 1.2 g (70%) of **15k** as a waxy solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta = 1.14-1.72$  (m, 32 H,  $16 \times \text{CH}_2$ ), 1.84–1.97 (m, 2 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.54 (t, J = 7.5 Hz, 2 H, 3-C $H_2$ ), 2.80 (t, J = 7.5 Hz, 2 H, 3'-C $H_2$ ), 3.32–3.40 (m, 2 H,  $CH_2OH$ ), 4.32 (br. s, 1 H, OH; interchangeable with  $D_2O$ ), 4.58 (t,  $J = 7.2 \text{ Hz}, 2 \text{ H}, \text{ NC}H_2$ , 7.21 (d, J = 7.8 Hz, 1 H, H4), 7.33 (t, J= 7.0 Hz, 1 H, H5, 8.04-8.12 (m, 3 H, H2, H2' and H5'), 8.50 (d,J = 8.0 Hz, 1 H, H4'), 8.96 (d, J = 6.0 Hz, 1 H, H6'), 9.08 (s, 1 H, H8')H2') ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 25.3, 25.5, 28.27, 28.30, 28.33, 28.70, 28.72, 28.80, 28.84, 28.86, 28.90 and 29.0  $(13 \times CH_2)$ , 29.7  $(3'-CH_2CH_2)$ , 29.9  $(3-CH_2CH_2)$ , 30.6  $(CH_2CH_2N)$ , 31.6 (3'- $CH_2$  + 3- $CH_2$ ), 32.5 ( $CH_2CH_2OH$ ), 60.7 (CH<sub>2</sub>OH, CH<sub>2</sub>N), 125.6 (C4), 126.0 (C5), 127.5 (C5'), 136.2 (C6), 138.0 (C2), 141.5 (C3'), 142.2 (C6'), 143.0 (C3), 143.9 (C2'), 145.1 (C4') ppm. IR (KBr):  $\tilde{v} = 3451 \text{ cm}^{-1} \text{ s}$ , 3219 s, 2922 s, 2850 s, 1602 m, 1563 w, 1466 m, 1437 m, 1343 w, 1268 m, 1154 s, 1058 m, 1027 m, 965 m, 874 w, 812 m, 759 m, 721 w, 681 m. HRMS: calcd. for  $C_{31}H_{51}N_2O_2^+$  483.3945; found 483.3939.

Cyclostellettamine H (1): A suspension of 15f (1.0 g, 1.6 mmol) in 15 mL of CHCl<sub>3</sub> under argon at 0 °C was treated with PBr<sub>3</sub> (0.6 mL, 6.4 mmol) over a period of 10 min. After stirring at 0 °C for an additional 15 min the mixture was heated to reflux for 1 h. It was then cooled to room temperature, poured into an ice-water solution (30 mL) and stirred until the ice had melted. The layers were separated and the aqueous layer was extracted three times with 20 mL of CH<sub>2</sub>Cl<sub>2</sub>. The combined organic phases were neutralised by washing three times with 20 mL of saturated NaHCO3 solution and dried with MgSO<sub>4</sub>. Most of the solvent was removed, leaving about 2 mL of bromide-containing solution, which was diluted with 13 mL of 2-butanone. The resulting mixture was added to a refluxing solution of NaI (0.53 g, 3.5 mmol) in 250 mL of 2butanone with a syringe pump at a rate of approximately 0.6 mL per hour. Refluxing was continued for 4 d. The solvent was then removed to yield a brown oil, which solidified upon trituration with Et<sub>2</sub>O. Recrystallisation from methanol/Et<sub>2</sub>O yielded 810 mg (70%) of 1 as an orange solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.00-1.37 (m, 28 H,  $14\times CH_2$ ), 1.54-1.74 (m, 4 H,  $3-CH_2CH_2$ ), 1.81-2.00 (m, 4 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.73-2.89 (m, 4 H, 3-CH<sub>2</sub>), 4.59 (t, J = 6.3 Hz, 4 H, NC $H_2$ ), 8.10 (dd, J = 6.1, J = 7.8 Hz, 2 H, H5), 8.49 (d, J = 7.5 Hz, 2 H, H4), 8.97 (d, J = 6.0 Hz, 2 H, H6), 9.09 (s, 2 H, H2) ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 24.8, 25.0, 27.4, 27.6, 28.0 28.3, 28.4, 28.5, 28.7 and 28.9 (10 × CH<sub>2</sub>) 29.3 and 29.6 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.2 and 30.4 (NCH<sub>2</sub>CH<sub>2</sub>), 31.2 and 31.3 (3-CH<sub>2</sub>), 60.4 (NCH<sub>2</sub>), 127.7 (C5), 142.4 (C6), 142.7 (C3), 144.0 (C2), 145.4 (C4) ppm. IR (KBr):  $\tilde{v} = 3014 \text{ cm}^{-1} \text{ m}$ , 2921 s, 2850 s, 1625 s, 1582 w, 1500 s, 1467 s, 1438 s, 1367 w, 1305 w, 1233 w, 1207 w, 1153 m, 1106 w, 1007 w, 926 w, 813 m, 765 w, 721 m, 696 s. HRMS: calcd. for C<sub>32</sub>H<sub>52</sub>IN<sub>2</sub>+ 591.3170; found 591.3155. calcd. for  $C_{32}H_{52}N_2^{2+}$  232.2060 (found 232.2061),  $C_{17}H_{28}N^+$  $246.2216\ (246.2217),\ C_{15}H_{24}N^{+}\ 218.1903\ (218.1919).$ 

Cyclostellettamine I (3): Prepared from 15g (1 g, 1.56 mmol), PBr<sub>3</sub> (0.58 mL, 6.3 mmol) and NaI (0.47 g, 3.12 mmol) as described above for 1. Compound 3 (0.75 g, 66%) was isolated as an offwhite solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta = 1.04-1.31$  (m, 30 H, 15×CH<sub>2</sub>), 1.57–1.70 (m, 4 H, 3-CH<sub>2</sub>CH<sub>2</sub>), 1.82–1.99 (m, 4 H, NCH<sub>2</sub>C $H_2$ ), 2.81 (t, J = 6.0 Hz, 4 H, 3-C $H_2$ ), 4.59 (t, J = 6.2 Hz, 4 H, NC $H_2$ ), 8.09 (pseudo t, J = 6.9 Hz, 2 H, H5), 8.50 (d, J =7.8 Hz, 2 H, H4), 8.96 (d, J = 5.5 Hz, 2 H, H6), 9.08 (s, 2 H, H2) ppm. <sup>13</sup>C NMR (62.5 MHz,  $[D_6]DMSO$ ):  $\delta = 24.9, 25.2, 27.6, 27.9,$ 28.3, 28.5, 28.7 and 28.9 (8 × CH<sub>2</sub>), 29.3 and 29.8 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.1 and 30.5 (NCH<sub>2</sub>CH<sub>2</sub>), 31.0 and 31.5 (3-CH<sub>2</sub>), 60.6 (NCH<sub>2</sub>), 127.7 (C5), 142.3 (C6), 142.8 (C3), 143.9 (C2), 145.4 (C4) ppm. IR (KBr):  $\tilde{v} = 3011 \text{ cm}^{-1} \text{ m}, 2921 \text{ s}, 2851 \text{ s}, 1625 \text{ m}, 1583 \text{ w}, 1500 \text{ s}, 1467 \text{ m},$ 1439 w, 1364 w, 1326 w, 1240 w, 1208 w, 1152 w, 924 w, 835 m, 801 w, 722 w, 696 s. HRMS: calcd. for  $C_{33}H_{54}IN_2^+$  605.3363 (found 605.3308),  $C_{33}H_{54}N_2^{2+}$  239.2138 (239.2136),  $C_{18}H_{30}N^+$  260.2373 (260.2364),  $C_{15}H_{24}N^+$  218.1903 (218.1914).

Cyclostellettamine K (4): Prepared from 15h (1 g, 1.53 mmol), PBr<sub>3</sub> (0.57 mL, 6.2 mmol) and NaI (0.51 g, 3.36 mmol) as described above for 1. Compound 4 (0.81 g, 71%) was isolated as a slightly yellow solid.  $^{1}$ H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.01–1.32 (m, 32 H,  $16 \times \text{CH}_2$ ), 1.55 – 1.74 (m, 4 H,  $3 - \text{CH}_2\text{C}H_2$ ), 1.83 – 1.99 (m, 4 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.74–2.88 (m, 4 H, 3-CH<sub>2</sub>), 4.60 (t, J = 5.8 Hz, 4 H, NC $H_2$ ), 8.10 (dd, J = 6.0, J = 7.9 Hz, 2 H, H5), 8.50 (d, J =8.0 Hz, 2 H, 4 H, 8.99 (d, J = 5.8 Hz, 2 H, 4 H6), 9.10 (s, 2 H, 4 H2) ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 24.6, 25.1, 27.2, 27.9, 28.0, 28.3, 28.5, 28.6, 28.8 and 28.9 (10×CH<sub>2</sub>), 29.0 and 29.7 (3-CH<sub>2</sub>CH<sub>2</sub>), 29.9 and 30.4 (NCH<sub>2</sub>CH<sub>2</sub>), 31.0 and 31.4 (3-CH<sub>2</sub>), 60.4 (NCH<sub>2</sub>), 127.7 (C5), 142.2 (C6), 142.6 (C3), 143.8 (C2), 145.2 (C4) ppm. IR (KBr):  $\tilde{v} = 3023 \text{ cm}^{-1} \text{ m}$ , 2921 s, 2850 s, 1718 w, 1629 m, 1500 s, 1466 s, 1364 w, 1321 w, 1238 w, 1202 w, 1150 w, 1029 w, 905 w, 827 w, 721 w, 688 s. HRMS: calcd. for  $C_{34}H_{56}N_2^{2+}$  246.2216 (found 246.2195),  $C_{19}H_{32}N^+$  274.2529 (274.2503),  $C_{15}H_{24}N^+$ 218.1903 (218.1912).

Cyclostellettamine G (2): Prepared from 15i (1 g, 1.56 mmol), PBr<sub>3</sub> (0.58 mL, 6.3 mmol) and NaI (0.47 g, 3.12 mmol) as described above for 1. Compound 2 (0.819 g, 72%) was isolated as a lightbrown solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta = 1.02-1.38$  (m, 30 H, 15×CH<sub>2</sub>), 1.55–1.74 (m, 4 H, 3-CH<sub>2</sub>CH<sub>2</sub>), 1.82–2.00 (m, 4 H, NCH<sub>2</sub>C $H_2$ ), 2.81 (t, J = 6.1 Hz, 4 H, 3-C $H_2$ ), 4.63 (t, J = 6.1 Hz, 4 H, NC $H_2$ ), 8.1 (dd, J = 6.1, J = 7.7 Hz, 2 H, H5), 8.51 (d, J =7.9 Hz, 2 H, H4), 9.03 (d, J = 5.9 Hz, 2 H, H6), 9.1 (s, 2 H, H2) ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 24.7, 24.9, 27.3, 27.7, 28.0, 28.3, 28.4, 28.6, 28.70, 28.75 and 28.80 (11×CH<sub>2</sub>), 29.2 and 29.4 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.1 and 30.2 (NCH<sub>2</sub>CH<sub>2</sub>), 31.1 and 31.4 (3-CH2), 60.4 (NCH<sub>2</sub>), 127.6 (C5), 142.2 (C6), 142.6 (C3), 143.9 (C2), 145.2 (C4) ppm. IR (KBr):  $\tilde{v} = 3013 \text{ cm}^{-1} \text{ m}$ , 2921 s, 2851 s,1719 w, 1626 m, 1501 s, 1467 s, 1444 m, 1364 w, 1314 w, 1238 w, 1206 w, 1153 w, 924 w, 816 m, 697 s. HRMS: calcd. for  $C_{33}H_{54}IN_2^+$ 605.3363 (found 605.3317),  $C_{33}H_{54}N_2^{2+}$  239.2138 (239.2135),  $C_{17}H_{28}N^+$  246.2216 (246.2211),  $C_{16}H_{26}N^+$  232.2060 (232.2064).

**Cyclostellettamine L (5):** Prepared from **15j** (0.8 g, 1.29 mmol), PBr<sub>3</sub> (0.45 mL, 4.8 mmol) and NaI (0.4 g, 2.64 mmol) as described above for **1**. Compound **5** (0.65 g, 66%) was isolated as a yellow solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 1.02–1.37 (m, 34 H, 17 × CH<sub>2</sub>), 1.56–1.74 (m, 4 H, 3-CH<sub>2</sub>CH<sub>2</sub>), 1.82–1.99 (m, 4 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.73–2.88 (m, 4 H, 3-CH<sub>2</sub>), 4.59 (t, J = 6.3 Hz, 4 H, NCH<sub>2</sub>), 8.11 (dd, J = 6.0, J = 7.7 Hz, 2 H, H5), 8.50 (d, J = 7.8 Hz, 2 H, H4), 8.97 (d, J = 5.7 Hz, 2 H, H6), 9.08 (s, 2 H, H2) ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 24.7, 25.0, 27.3, 27.7, 28.0, 28.3, 28.45, 28.50, 28.65, 28.70, 28.77, 28.80 and 28.88 (13 × CH<sub>2</sub>), 29.1 and 29.4 (3-CH<sub>2</sub>CH<sub>2</sub>), 29.9 and 30.2 (NCH<sub>2</sub>CH<sub>2</sub>), 31.2 and

31.3 (3-CH<sub>2</sub>), 60.4 (N–CH<sub>2</sub>), 127.6 (C5), 142.2 (C6), 142.6 (C3), 143.8 (C2), 145.3 (C4) ppm. IR (KBr):  $\tilde{v}=3016~\text{cm}^{-1}$  m, 2921 s, 2851 s, 1727 w, 1627 m, 1502 s, 1466 s, 1362 w, 1320 w, 1237 w, 1204 w, 1152 w, 1111 w, 1030 w, 927 w, 818 m, 720 m, 693 s. HRMS: calcd. for  $C_{35}H_{58}N_2^{2+}$  253.2295 (found 253.2264),  $C_{19}H_{32}N^+$  274.2529 (274.2499),  $C_{16}H_{26}N^+$  232.2060 (232.2049).

Cyclostellettamine Q (6): Prepared from 15k (1.27 g, 2.08 mmol), PBr<sub>3</sub> (0.78 mL, 8.3 mmol) and NaI (0.69 g, 4.58 mmol) as described above for 1. Compound 6 (0.59 g, 40%) was isolated as an off-white solid. <sup>1</sup>H NMR (250 MHz, [D<sub>6</sub>]DMSO):  $\delta = 0.99-1.30$ (m, 26 H,  $13 \times \text{CH}_2$ ), 1.55-1.71 (m, 4 H,  $3-\text{CH}_2\text{C}H_2$ ), 1.82-1.98 (m, 4 H, NCH<sub>2</sub>CH<sub>2</sub>), 2.80 (t, J = 6.3 Hz, 4 H, 3-CH<sub>2</sub>), 4.60 (t, J =5.9 Hz, 4 H, NC $H_2$ ), 8.10 (t, J = 6.9 Hz, 2 H, H5), 8.50 (d, J =8.0 Hz, 2 H, H4), 9.05 (d, J = 5.8 Hz, 2 H, H6), 9.20 (s, 2 H, H2)ppm. <sup>13</sup>C NMR (62.5 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 24.9, 25.0, 27.5, 27.7, 28.0, 28.2, 28.4, 28.5, 28.6, 28.7, 28.8, 28.9 and 29.0  $(13 \times CH_2)$ , 29.4 and 29.6 (3-CH<sub>2</sub>CH<sub>2</sub>), 30.2 and 30.4 (NCH<sub>2</sub>CH<sub>2</sub>), 31.2 and 31.4 (3-CH<sub>2</sub>), 60.4 (NCH<sub>2</sub>), 127.7 (C5), 142.4 (C6), 142.7 (C3), 144.0 (C2), 145.3 (C4) ppm. IR (KBr):  $\tilde{v} = 3013 \text{ cm}^{-1} \text{ m}$ , 2923 s, 2852 s, 1627 m, 1583 w, 1500 s, 1467 m, 1439 w, 1368 w, 1321 w, 1241 w, 1205 w, 1154 w, 823 m, 758 w, 724 w, 697 m. HRMS: calcd. for  $C_{31}H_{50}IN_2^+$  577.3019 (found 577.3012),  $C_{31}H_{50}N_2^{2+}$  225.1982 (225.1991).

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- 1993, 58, 5925–5930; e) K. Sepčić, G. Guella, I. Mancini, F. Pietra, M. Dalla Serra, G. Menestrina, K. Tubbs, P. Maček, T. Turk, *J. Nat. Prod.* 1997, 60, 991–996; f) C. A. Volk, M. Köck, *Org. Lett.* 2003, 5, 3567–3569; g) C. A. Volk, M. Köck, *Org. Biomol. Chem.* 2004, 2, 1827–1830.
- [3] F. J. Schmitz, K. H. Hollenbeak, D. C. Campbell, J. Org. Chem. 1978, 43, 3916–3922.
- [4] a) N. Fusetani, K. Yasumuro, S. Matsunaga, H. Hirota, *Tetrahedron Lett.* 1989, 30, 6891–6894; b) C. A. Volk, H. Lippert, E. Lichte, M. Köck, *Eur. J. Org. Chem.* 2004, 3154–3158.
- [5] a) N. Fusetani, N. Asai, S. Matsunaga, K. Honda, K. Yasumuro, *Tetrahedron Lett.* 1994, 35, 3967–3970; b) N. Oku, K. Nagai, N. Shindoh, Y. Terada, R. W. M. van Soest, S. Matsunaga, N. Fusetani, *Bioorg. Med. Chem. Lett.* 2004, 14, 2617–2620; c) J. H. H. L. de Oliveira, A. Grube, M. Köck, R. G. S. Berlinck, M. L. Macedo, A. G. Ferreira, E. Hajdu, *J. Nat. Prod.* 2004, 67, 1685–1689.
- [6] A 1.5 mg sample of the mixture was obtained from this sponge and analysed. The 1D <sup>1</sup>H NMR spectrum of the mixture showed no indication of compounds other than the cyclostellettamines. The compounds were identified by their characteristic UV maximum at a wavelength of 267 nm, the measured high resolution mass and their MS fragmentation pattern.
- [7] Due to an overlap in publishing the results with Fusetani's work (ref.<sup>[5b]</sup>) the name cyclostellettamine G has already been given to the chain length combination 11 and 12, which would have been cyclostellettamine H in our nomenclature.
- [8] This compound is a synthetic precursor of haliclamine D and has not yet been isolated as a natural product. The name cyclostellettamine Q was chosen because it is probable that further derivatives will be isolated in the future. Cyclostellettamines M, N, O, P and R are reserved for the chain length combinations 9/9, 9/10, 10/10, 9/11 and 11/11, respectively.
- [9] J. E. Baldwin, D. R. Spring, C. E. Atkinson, V. Lee, *Tetrahedron* 1998, 54, 13655–13680.
- [10] S. K. Kang, W. S. Kim, B. H. Moon, Synthesis 1985, 1161– 1162.
- [11] a) H. J. Veith, J. H. Gross, Org. Mass Spectrom. 1991, 26, 1061–1064; b) H. J. Veith, J. H. Gross, Org. Mass Spectrom. 1991, 26, 1097–1108.
- [12] A. M. Duffield, R. Beugelmans, H. Budzikiewicz, D. A. Lightner, D. H. Williams, C. Djerassi, J. Am. Chem. Soc. 1965, 87, 805–810.
- [13] G. Spiteller, Adv. Heterocycl. Chem. 1966, 7, 301-376.
- [14] K. Mori, N. P. Argade, Liebigs Ann. Chem. 1994, 695-700.
- [15] M. J. Wanner, G.-J. Koomen, Eur. J. Org. Chem. 1998, 889– 895.

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R. J. Andersen, R. W. M. van Soest, F. Kong, *Alkaloids: Chemical and Biological Perspectives*, S. W. Pelletier: Pergamon/Elsevier, Oxford, 1996, vol. 10, pp. 301–355.

<sup>[2]</sup> a) G. Cimino, S. de Stefano, G. Scognamiglio, G. Sodano, E. Trivellone, Bull. Soc. Chim. Belg. 1986, 95, 783–800; b) R. Sakai, T. Higa, C. W. Jefford, G. Bernardinelli, J. Am. Chem. Soc. 1986, 108, 6404–6405; c) R. Sakai, S. Komoto, T. Higa, C. W. Jefford, G. Bernardinelli, Tetrahedron Lett. 1987, 28, 5493–5496; d) M. T. Davies-Coleman, D. J. Faulkner, G. M. Dubowchik, G. P. Roth, C. Polson, C. Fairchild, J. Org. Chem.